Physico-Chemical Properties of Fluorinated Oligo(azomethine)s with Azo Groups in the Main Chain



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Abstract: We report the synthesis and optical behavior of photoactive fluorinated azo-based oligo(azomethine)s (Azo-Oam). The desired oligomers were prepared by polycondensation of octafluorobiphenylene-containing diamine with excess of tetrafluorobenzene- or octafluorobiphenylene-based bis-hydroxybenzaldehydes. The repeating unit of the oligomers was targeted as n=7 by controlling the molar ratio of initial monomers. Importantly, the oligomers can be solution cast into flexible solid films with tensile strength in the range 13-20 MPa. The studied Azo-Oam showed a remarkable response to both optical and chemical stimuli. Thus, the trans-cis photoisomerization of azobenzene units occurs in Azo-Oam solid films as well as absorption maxima of the obtained oligomers can be regulated by changing the pH of a medium and a solvent concentration. It was studied that the irradiation of the synthesized oligomers leads to the emergence of birefringence in their films. The highly stable diffraction gratings based on Azo-Oam's films were fabricated which can be stored for the long time.

Advantages of combining azo and azomethine groups in the molecule conjugation system The advantages of fluorine-containing polymers

- -N=N-+-C=N-H-C=N--N=N-✓ Good distribution of electron ✓ Liquid-crystal properties ✓ High optical sensitivity density ✓ Piezo and pyroelectric ✓ Bathochromic shift ✓ Nonlinear optical properties properties \checkmark Good complexing ability ✓ Photo induced *trans-cis-*✓ Fiber formation ✓ Tautomeric transitions transitions ✓ Low dielectric anisotropy High thermal and thermooxidative stability The aim of this work is to study the optical properties of azocontaining olygoazomethines (Azo-Oam) with fluorinated fragments ✓ Improved chemical resistance
 - ✓ Low surface energy
- ✓ Low refractive index
- ✓ High optical transparency
- ✓ Good mechanical properties
- ✓ High electronegativity

polymer Azo-Oam-1

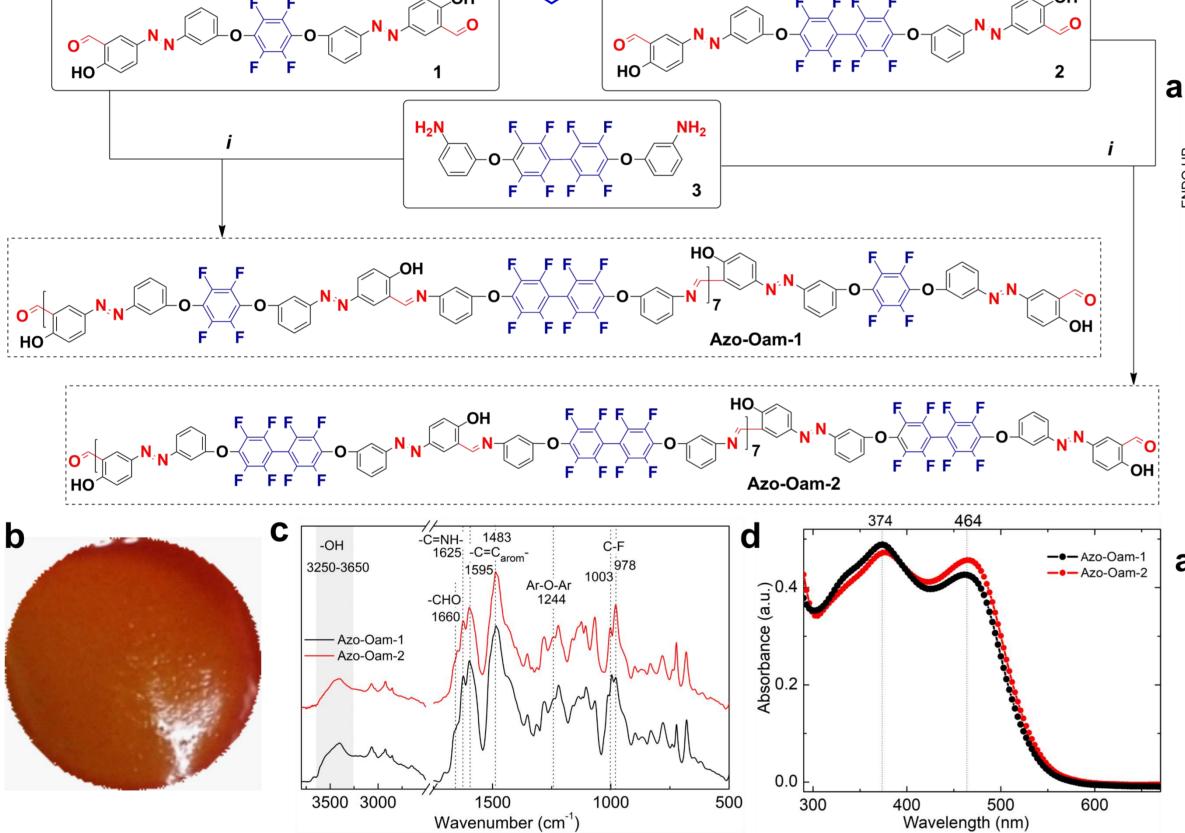


Figure 1. a) Synthesis of Azo-Oam-1 and Azo-Oam-2 oligomers: (i) DMAc, 110 °C, 24 h. Yield about 86% for both oligomers after reprecipitation. b) Digital image of self-standing Azo-Oam-2 oligomer (thickness ~100 µm). c) FTIR spectra of synthesized Azo-Oam oligomers; d) UVvis absorbance spectra of oligomers Azo-Oam-1 and Azo-Oam-2 in DMAc (0.006 mg/mL).

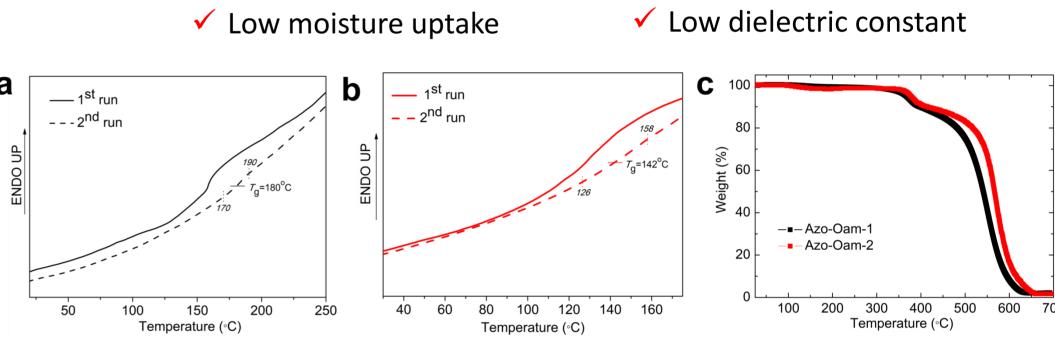


Figure 2. DSC traces (heat flow vs. temperature) of oligomers Azo-Oam-1 (a) and Azo-Oam-2 (b) recorded for film samples; depicted are the first heating and second heating runs. (DSC curves are offset for clarity). Air atmosphere, heating rate of 20°C/min. c) TGA curves of Azo-Oam-1 and Azo-Oam-2 in an air atmosphere (heating rate: 20 °C/min with a temperature from 25 to 700°C).

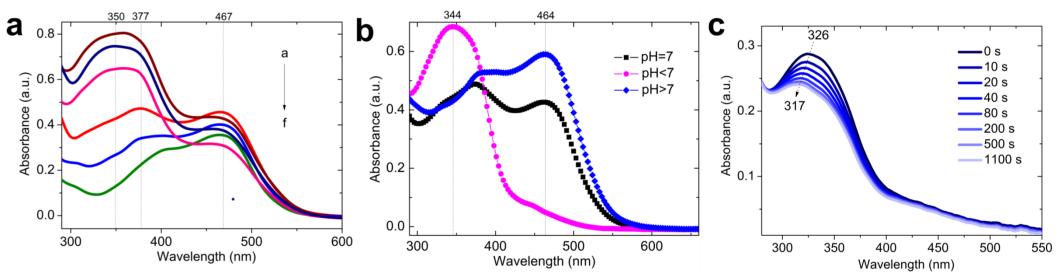
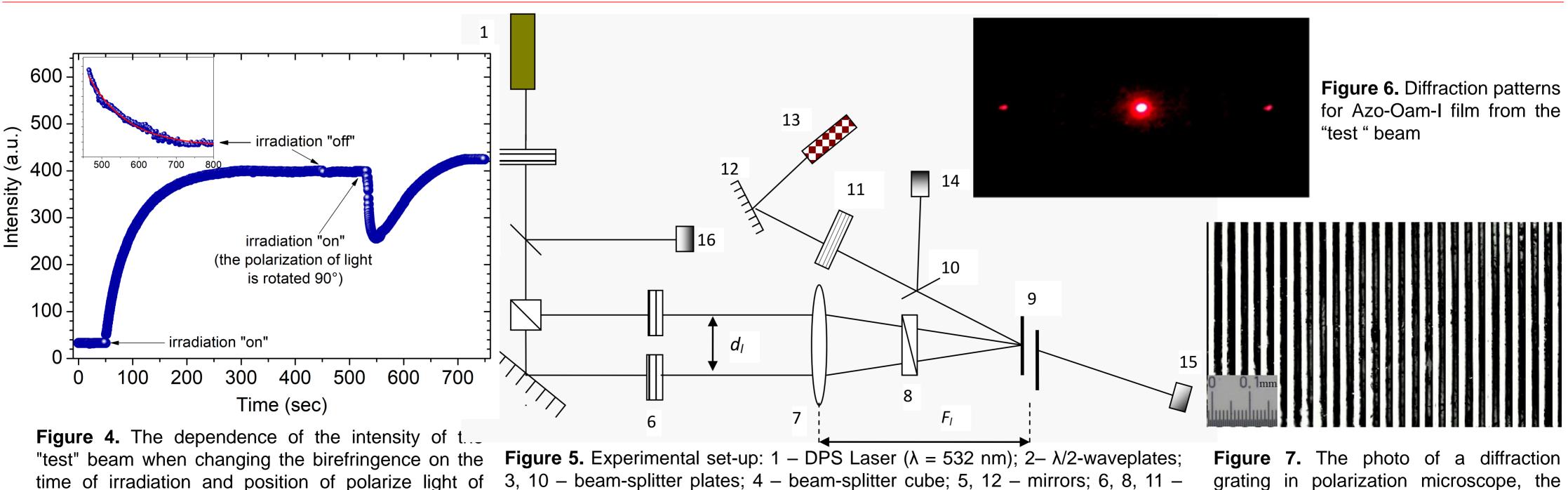


Figure 3. a) UV-vis spectra of Azo-Oam-2 at various concentrations (a-f) in DMAc. The concentration (a-f) of Azo-Oam-2 in DMAc solutions was about 15.0 (a), 13.8 (b), 11.0 (c), 6.0 (d), 4.5 (e) and 3.5 (f) µg mL⁻¹. b) UV-vis spectra of Azo-Oam-1 in DMAc at acidic (pH<7), neutral (pH=7) and alkaline medium (pH>7). c) Trans-cis photoisomerization of Azo-Oam-1 in thin film (λ =370 nm, P 3.4 mW)



film of Azo-Oam-1 (scale 0,1 mm)

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photodiodes.

polarizer; 7 – lens; 9 – sample; 13 – He-Ne laser (λ = 632.8 nm); 14, 15, 16 –